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Chapter 02 - The Chemical Context of Life | CourseNotes

Introduction. Dendrimers are nano-sized, radially symmetric molecules with well-defined, homogeneous, and monodisperse structure consisting of tree-like arms or branches [1]. These hyperbranched molecules were first discovered by Fritz Vogtle in 1978, by Donald Tomalia and co-workers in the early 1980s, and at the same time, but independently by George R. Newkome.

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Section Review 2-1 1. protons; neutrons 2. electrons 3. neutrons 4. electrons 5. ionic 6. The two main types of chemical bonding are ionic and covalent bonding. Ionic bonds are formed when a

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Dendrimers: synthesis, applications, and properties

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Halogen bonding is a type of non-covalent interaction which does not involve the formation nor breaking of actual bonds, but rather is similar to the dipole-dipole interaction known as hydrogen bonding. In halogen bonding, a halogen atom acts as an electrophile, or electron-seeking species, and forms a weak electrostatic interaction with a nucleophile, or electron-rich species.

4 Consequences of Bullying Behavior. Bullying behavior is a serious problem among school-age children and adolescents; it has short- and long-term effects on the individual who is bullied, the individual who bullies, the individual who is bullied and bullies others, and the bystander present during the bullying event.

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A covalent bond, also called a molecular bond, is a chemical bond that involves the sharing of electron pairs between atom-

s. These electron pairs are known as shared pairs or bonding pairs, and the stable balance of attractive and repulsive forces between atoms, when they share electrons, is known as covalent bonding. For many molecules, the sharing of electrons allows each atom to attain the ...

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Pauli Exclusion Principle. The Pauli exclusion principle states that no two electrons can have the same four quantum numbers. The first three (n , l , and m_l) may be the same, but the fourth quantum number must be different. A single orbital can hold a maximum of two electrons, which must have opposing spins; otherwise they would have the same four quantum numbers, which is forbidden.

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Dendrimers: synthesis, applications, and properties

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